

BS/2111

CHEMISTRY-3-BHB-14

(Semester-III)

Time : Three Hours]

[Maximum Marks : 74

Note : Section A and Section-B consists of 4 questions and candidates are required to attempt any 2 questions and each will carry 11 marks. Section-C consists of 15 questions each will carry 2 marks.

SECTION-A

- I. (a) Explain the reduction of methods of ketones from acid chloride. (5)
- (b) Explain the three methods of formation of phenols. (6)
- II. (a) Out of alcohols and phenols which is more acidic and explain ? (6)
- (b) Explain the carboxylation. (5)
- III. (a) Explain the acidity of carboxylic acids. (6)
- (b) Explain the methods of formation of aldehydes. (5)

- IV. (a) Explain the effect of substitution on carboxylic acids. (6)
- (b) Explain the *two* reactions of formation of 1,3-dithianes. (5)

SECTION-B

- V. (a) Explain the First law of thermodynamics and its drawbacks. (6)
- (b) Explain the intensive and extensive properties. (5)
- VI. (a) Explain the different types of systems with examples. (6)
- (b) Explain the heat capacity at constant pressure. (5)
- VII. (a) Difference between state functions and path functions. (6)
- (b) Derive the relationship between C_v and C_p . (5)
- VIII. (a) Calculate the "q" for isothermal and adiabatic reversible process. (6)
- (b) Explain the formation of chemical reaction of amides. (5)

SECTION-C

- IX. (a) What is system and surroundings ?
- (b) Explain the 1,3 dithianes.
- (c) What is enthalpy ?

- (d) What are aldehyde and ketones ?
 - (e) A 0.5 mole of gas at temperature 300 K expands isothermally from an initial volume of 2 L to 6 L. What is the work done by the gas ?
 - (f) Explain the Joule Thomson effect.
 - (g) What are adiabatic process ?
 - (h) What is the heat ?
 - (i) What is the inversion temperature ?
 - (j) What type of hydrogen bonding?
 - (k) Explains the Joule law.
 - (l) What are exothermic and endothermic reactions ?
 - (m) What is the condition of internal energy for a spontaneous reaction ?
 - (n) Give *two* examples of open system.
 - (o) What is value of work done in a cyclic process ?
(15×2=30)
-